

## ON THE NATURE OF $P_i$ -INDUCED, $Mg^{2+}$ -PREVENTED $Ca^{2+}$ RELEASE IN RAT LIVER MITOCHONDRIA

Paolo BERNARDI and Daniela PIETROBON

*CNR Unit for the Study of Physiology of Mitochondria and Institute of General Pathology, Via Loredan 16, 35100 Padova, Italy*

Received 1 December 1981

### 1. Introduction

The pathway and role of mitochondrial  $Ca^{2+}$  efflux has assumed a major importance (review [1–3]). Although it is generally accepted that at steady state  $Ca^{2+}$  undergoes a continuous, slow recycling across the inner mitochondrial membrane, due to the simultaneous operation of the influx and efflux pathways [4], the nature and regulation of the efflux pathway are matters of debate [1–3].

Since 1964 it has been known that inorganic phosphate ( $P_i$ ) releases  $Ca^{2+}$  from liver mitochondria [5]. The  $Ca^{2+}$  release is irreversible and accompanied by indefinite stimulation of respiration [5]. Although these phenomena have been attributed to increased  $Ca^{2+}$  cycling, and therefore to stimulation by  $P_i$  of a specific  $Ca^{2+}$  efflux pathway [6–8], direct evidence for such a specific pathway and for real  $Ca^{2+}$  cycling has never been provided.

Here, we aim to clarify the nature of  $P_i$ -induced,  $Mg^{2+}$ -prevented  $Ca^{2+}$  release and of the related respiratory stimulation. We show that, in the absence of  $Mg^{2+}$  and BSA, phosphate addition to  $Ca^{2+}$ -preloaded mitochondria causes a sharp decrease of  $\Delta pH$ . Since  $\Delta\psi$  tends to decline as well, the  $H^+$  electrochemical gradient ( $\Delta\tilde{\mu}_{H^+}$ ) decreases dramatically. The nature of  $Ca^{2+}$  release induced by  $P_i$  in the absence of  $Mg^{2+}$  appears therefore to be due to reversal of the uniport

carrier, consequent to decrease of  $\Delta\psi$  rather than to activation of an independent efflux pathway. Accordingly, the indefinite stimulation of respiration is not due to increased  $Ca^{2+}$  cycling but to increased  $H^+$  recycling across the inner membrane.

### 2. Materials and methods

Rat liver mitochondria were prepared as in [9] in 0.25 M sucrose, 10 mM Tris-HCl (pH 7.4) and 0.1 mM EGTA. The final washing was carried out in an EGTA-free medium and mitochondrial protein was assayed with the biuret method, using BSA as a standard. The incubation media are specified in the figure legends.

Oxygen consumption was monitored with a Clark oxygen electrode (Yellow Springs Instruments, OH) in a magnetically stirred, thermo-equilibrated vessel.

$Ca^{2+}$  movements were monitored with a  $Ca^{2+}$ -selective electrode (W. Möller, Zürich) in a waterjacket-thermostatted vessel equipped with magnetic stirring. The  $Ca^{2+}$ -electrode was routinely calibrated with  $Ca^{2+}$ -HEDTA buffers containing 50 mM HEDTA, 2 mM  $MgCl_2$ , Tris-Mops buffer (pH 7.0) and  $CaCl_2$  to give the desired free  $[Ca^{2+}]$  (apparent stability constants were taken as  $2.57 \times 10^5$  ( $Ca^{2+}$ ) and  $1.12 \times 10^3$  ( $Mg^{2+}$ ) at pH 7.0 [4]). The value of  $[Ca-HEDTA_{total}]$  was taken as being equal to the total  $[Ca^{2+}]$ , since  $[Ca^{2+}] \ll [Ca-HEDTA_{total}]$ .

In the isotope experiments mitochondria were incubated under the specified conditions with  $[^{14}C]$ -TPMP<sup>+</sup> (0.02  $\mu Ci/ml$ , corresponding to final conc. 20  $\mu M$  TPMP<sup>+</sup>) or with  $[^{14}C]$ DMO (0.07  $\mu Ci/ml$ ). After the specified time the suspensions were centrifuged at  $30\,000 \times g$  in a Sorvall RC2B refrigerated

**Abbreviations:** EGTA, (ethylene-bis(oxoethylenitrilo))tetraacetic acid; BSA, bovine serum albumin; Tris, 2-amino-2-hydroxymethyl-1,3-propanediol; Mops, 4-morpholinepropane sulfonic acid; HEDTA, *N'*-(2-hydroxyethyl)ethylene-diamine-*N,N,N'*-triacetic acid; TPMP<sup>+</sup>, triphenylmethylphosphonium ion; DMO, 5,5-dimethyl-2,4-oxazolidinedione;  $pCa_o$ ,  $-\log [Ca^{2+}]$  outside the mitochondrial compartment;  $\Delta\psi$ , membrane potential

supercentrifuge for 5 min. The clear supernatants were decanted, the tube wall blotted dry, and the pellets dissolved by treatment with 0.2 ml of a 1 mM Na-EDTA solution containing 0.1% (w/w) NaCl and 0.9% (w/w) sodium deoxycholate, at room temperature. The dissolved pellets and 100  $\mu$ l aliquots of the supernatants were added to 2 ml of Packard Insta-Gel scintillation fluid. Mitochondrial volumes were calculated on parallel samples containing  $^3\text{H}_2\text{O}$  (1  $\mu\text{Ci}/\text{ml}$ ) and [ $^{14}\text{C}$ ]sucrose (0.2  $\mu\text{Ci}/\text{ml}$ ). The radioactivity was assayed with a Packard Tri Carb 300 C liquid scintillation spectrometer and the dpm were calculated using an external standard method and calibration curves to correct for quenching [10].

Membrane potential was calculated on [ $^{14}\text{C}$ ]TPMP $^+$  distribution across the inner mitochondrial membrane. A correction for passive TPMP $^+$  binding was introduced, measuring the amount of radioactivity present in mitochondria treated with FCCP + valinomycin + antimycin A in the presence 0–0.5 mM unlabeled TPMP $^+$ . Since the radioactivity found in the pellets did not decrease at still higher concentrations of TPMP $^+$ , the difference between the dpm at 0.5 mM and 0 TPMP $^+$  is taken as passive binding. This correction rests on the assumption that the TPMP $^+$  binding is the same in energized and de-energized mitochondria. The activity coefficient assumed for intramitochondrial TPMP $^+$  was 1. Obviously, both assumptions are only approximately correct, and the values of  $\Delta\psi$  could be overestimated. The membrane potential was calculated from the Nernst equation.

$\Delta\text{pH}$  was measured on [ $^{14}\text{C}$ ]DMO distribution across the inner membrane, according to [11].

### 3. Results

Fig.1A shows the effects of  $\text{P}_i$  on steady state  $\text{Ca}^{2+}$  distribution in mitochondria incubated in the absence of  $\text{Mg}^{2+}$  and BSA. After the uptake of  $\sim 15$  nmol  $\text{Ca}^{2+}/\text{mg}$  protein, addition of 2 mM  $\text{P}_i$  caused only a very slight increase of  $\text{Ca}^{2+}$  uptake. When a further 80  $\mu\text{M}$   $\text{Ca}^{2+}$  was added,  $\text{Ca}^{2+}$  uptake was followed by a sudden and irreversible  $\text{Ca}^{2+}$  release. Thus, in the absence of  $\text{Mg}^{2+}$ ,  $\text{P}_i$  does not significantly alter the steady state  $\text{Ca}^{2+}$  distribution while it induces  $\text{Ca}^{2+}$  efflux at high  $\text{Ca}^{2+}$  loads. Fig.1B shows that, in the presence of  $\text{Mg}^{2+}$  and BSA, the steady state extramitochondrial  $\text{pCa}_0$  is poised at lower levels, in accord with [4,12]. Addition of  $\text{P}_i$  then caused a slow  $\text{Ca}^{2+}$  efflux until a higher steady state  $\text{pCa}_0$  was attained.

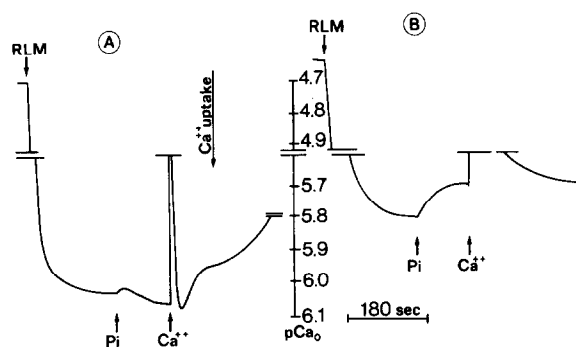


Fig.1. Prevention by  $\text{Mg}^{2+}$  and BSA of  $\text{P}_i$ -induced  $\text{Ca}^{2+}$  release. (A) Incubation medium contained 0.14 M sucrose, 40 mM choline chloride, 10 mM Tris-Mops (pH 7.0), 5 mM succinate-Tris, 16  $\mu\text{M}$   $\text{Ca}^{2+}$ . In (B) 2 mM  $\text{MgCl}_2$  and 1 mg/ml BSA were added. Final volume 5 ml, 30°C. When indicated 5 mg mitochondria (RLM), 2 mM  $\text{P}_i$  and 80  $\mu\text{M}$   $\text{Ca}^{2+}$ .

The specific features of this novel  $\text{P}_i$ -stimulated  $\text{Ca}^{2+}$  efflux are analyzed in detail in [13]. It is important here to stress that in the presence of BSA and  $\text{Mg}^{2+}$  a further 80  $\mu\text{M}$   $\text{Ca}^{2+}$  pulse may be taken up, and that the same steady state  $\text{Ca}^{2+}$  distribution preceding the first  $\text{Ca}^{2+}$  addition is reached and maintained until anaerobiosis (not shown). Thus the process of complete and irreversible  $\text{Ca}^{2+}$  release caused by  $\text{P}_i$  is abolished by BSA and  $\text{Mg}^{2+}$ .

Fig.2 analyzes the nature of the respiratory stimu-

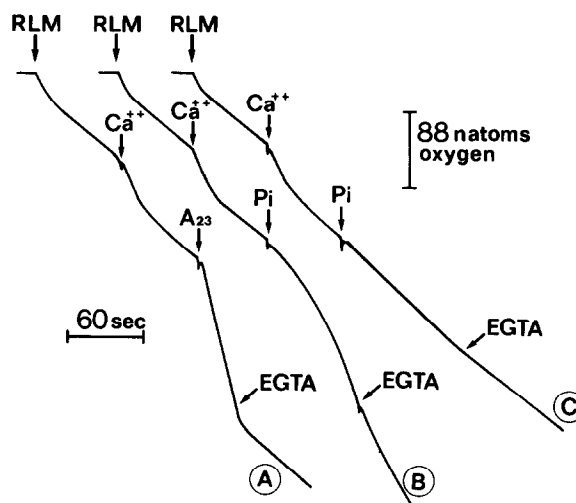


Fig.2. Effect of BSA,  $\text{Mg}^{2+}$  and EGTA on respiratory stimulation following  $\text{Ca}^{2+}$  and  $\text{P}_i$  uptake. The incubation medium was the same as fig.1A plus 5 mM acetate-Tris. In (C) 2 mM  $\text{MgCl}_2$  and 1 mg/ml BSA were added. Final volume 2 ml, 30°C. When indicated, 4 mg mitochondria (RLM), 90  $\mu\text{M}$   $\text{Ca}^{2+}$ , 2  $\mu\text{g}$  A23187 (A 23), 2 mM  $\text{P}_i$  and 0.25 EGTA.

lation due to  $\text{Ca}^{2+}$  plus  $\text{P}_i$ . Trace 2A shows that addition of the electroneutral  $\text{Ca}^{2+}$  ionophore A23187 after a  $\text{Ca}^{2+}$  pulse caused an immediate increase of respiration, due to continuous  $\text{Ca}^{2+}$  recycling across the inner membrane [14]. This is confirmed by the fact that addition of EGTA brought the respiration to the state 4 level preceding the addition of the ionophore. In trace 2B, after the  $\text{Ca}^{2+}$  pulse, 2 mM  $\text{P}_i$  was added instead of A23187. Respiration was gradually stimulated after a lag phase [5,7,8] while addition of EGTA had only a very slight effect. Trace 2C indicates that in the presence of  $\text{Mg}^{2+}$  and BSA,  $\text{P}_i$  had negligible effects on mitochondrial respiration, and that EGTA again slightly depressed state 4 respiration.

Table 1 analyzes the values of  $\Delta\psi$ ,  $\Delta\text{pH}$  and  $\Delta\tilde{\mu}_{\text{H}^+}$  in  $\text{Ca}^{2+}$ -loaded mitochondria, and the effects of  $\text{P}_i$  on these values in the presence or absence of  $\text{Mg}^{2+}$  plus BSA.

In the absence of  $\text{P}_i$ ,  $\text{Ca}^{2+}$ -loaded mitochondria maintained a  $\Delta\text{pH}$  of 89 mV and a  $\Delta\psi$  of 121 mV, with a total  $\Delta\tilde{\mu}_{\text{H}^+}$  of 210 mV.  $\Delta\text{pH}$  was decreased to very low values after addition of  $\text{P}_i$ .  $\Delta\psi$  on the other hand increased markedly in the presence but not in the absence of  $\text{Mg}^{2+}$  + BSA, almost completely compensating the drop in  $\Delta\text{pH}$  in the former but not in the latter case. Furthermore  $\Delta\psi$  was kept at a higher level after addition of EGTA (101 mV) but not after

addition of ruthenium red. Table 1 thus shows that partial restoration of  $\Delta\tilde{\mu}_{\text{H}^+}$  requires removal of  $\text{Ca}^{2+}$  from the medium and not simple abolition of  $\text{Ca}^{2+}$  cycling by ruthenium red.

#### 4. Discussion

Since the early fifties it has been known that addition of  $\text{P}_i$  results in a large-amplitude mitochondrial swelling which is reversed by  $\text{ATP} + \text{Mg}^{2+}$  (review [15,16]). The  $\text{P}_i$ -induced swelling is markedly enhanced by the simultaneous uptake of  $\text{Ca}^{2+}$ , and is accompanied by an increased permeability of the mitochondrial membrane and by subsequent release of matrix constituents [17,18]. In 1964 it was shown that  $\text{P}_i$  releases  $\text{Ca}^{2+}$  from mitochondria [5].  $\text{Ca}^{2+}$  release required a threshold concentration of  $\text{Ca}^{2+}$  or  $\text{P}_i$ , and was accompanied by an indefinite stimulation of the respiration, which was attributed to increased  $\text{Ca}^{2+}$  cycling across the inner membrane [5]. This view has been re-evaluated by different groups [6–8], and it has been tacitly assumed that stimulation of respiration after completion of  $\text{Ca}^{2+}$  uptake is a reflection of increased  $\text{Ca}^{2+}$  cycling [6–8]. Due to this view, it has been concluded that  $\text{Sr}^{2+}$ , at variance from  $\text{Ca}^{2+}$ , is unable to cycle across the mitochondrial membrane from the observation that addition of  $\text{P}_i$  after  $\text{Sr}^{2+}$  uptake does not result in stimulation of respiration [7].

Our data indicate that the  $\text{P}_i$ -induced,  $\text{Mg}^{2+}$ - and BSA-prevented  $\text{Ca}^{2+}$  efflux is due to a reversal of  $\text{Ca}^{2+}$  transport through the uniport carrier following a decrease of the transmembrane electrical potential. Furthermore, the related increase in respiration is mainly due to an increased  $\text{H}^+$  recycling and not to a  $\text{Ca}^{2+}$  recycling. This is supported by four lines of evidence:

- (i) The  $\text{P}_i$ -induced  $\text{Ca}^{2+}$  efflux occurring in the absence of  $\text{Mg}^{2+}$  (fig.1A) and the related increase of respiration (fig.2) are prevented by  $\text{Mg}^{2+}$  and BSA, which are agents well known to protect against membrane damage [19];
- (ii) The  $\text{P}_i$ -induced  $\text{Ca}^{2+}$  efflux does not lead to a new set point for  $\text{Ca}^{2+}$  distribution, as required by the establishment of a kinetic steady state between  $\text{Ca}^{2+}$  influx and efflux processes, but rather to a complete  $\text{Ca}^{2+}$  release;
- (iii) The stimulation of respiration induced by  $\text{Ca}^{2+}$  plus  $\text{P}_i$  is almost unaffected by the addition of EGTA, which is expected to stop  $\text{Ca}^{2+}$  cycling, as it occurs in the case of the ionophore A23187 (fig.2A);

Table 1  
 $\Delta\psi$  and  $\Delta\text{pH}$  following  $\text{Ca}^{2+}$  and  $\text{P}_i$  uptake

Exp.	Addition(s) to basic medium	$\Delta\psi$ (mV)	$\Delta\text{pH}$ (mV)	$\Delta\tilde{\mu}_{\text{H}^+}$ (mV)
1	None	121	89	210
2	2 mM $\text{P}_i$	47	0	47
3	2 mM $\text{P}_i$ , 1 mg/ml BSA, 2 mM $\text{Mg}^{2+}$	164	10	174
4	2 mM $\text{P}_i$ , 2 $\mu\text{M}$ ruthenium red	56	4	60
5	2 mM $\text{P}_i$ , 0.5 mM EGTA	101	10	111

The incubation medium was as in fig.1A plus 20  $\mu\text{M}$  [ $^{14}\text{C}$ ]-TPMP\* ( $\Delta\psi$  measurements) or carrier-free [ $^{14}\text{C}$ ]DMO ( $\Delta\text{pH}$  measurements) in parallel samples. Total calcium was 80 nmol/mg protein. Further additions to the medium were as indicated. The experiments were started by the addition of 5 mg mitochondria to 2 ml final vol. at room temperature. In exp. 1–3  $\Delta\psi$  and  $\Delta\text{pH}$  were determined after 4 min incubation. In exp. 4,5, ruthenium red or EGTA were added after 4 min incubation, and  $\Delta\psi$  and  $\Delta\text{pH}$  were determined after a further 1 min. Matrix volumes were calculated on parallel samples containing  $^3\text{H}_2\text{O}$  and [ $^{14}\text{C}$ ]sucrose. For further explanation see section 2

(iv) The data of table 1 indicate that in the absence of  $P_i$ ,  $Ca^{2+}$  uptake results in establishment of a large  $\Delta pH$ .

In the absence of  $Mg^{2+}$  and BSA, addition of  $P_i$  causes a collapse of  $\Delta pH$  without parallel increase of  $\Delta\psi$ . The lack of  $\Delta\psi$  rise, to an extent complementary to the depression of  $\Delta pH$ , as required to maintain an unaltered  $\Delta\tilde{\mu}_{H^+}$ , cannot be ascribed to energy-dissipating  $Ca^{2+}$  recycling, since ruthenium red does not cause increase of  $\Delta\psi$ . The partial  $\Delta\psi$  recovery induced by EGTA can be explained by inhibition of  $Ca^{2+}$ -stimulated phospholipases [20]. Thus the finding of a 5 mV decrease of the membrane potential upon addition of 1.6 mM  $P_i$ , with concomitant  $Ca^{2+}$  efflux [6], means in fact a presumable 60 mV decrease of  $\Delta\tilde{\mu}_{H^+}$ . Furthermore, when  $\Delta\psi$  is below the critical value of  $\sim 125$  mV,  $Ca^{2+}$  distribution is at electrochemical equilibrium, and any decrease of  $\Delta\psi$  allows  $Ca^{2+}$  efflux to occur via a reversal of the uniport  $Ca^{2+}$  carrier [4].

In conclusion we suggest that distinction between a  $Ca^{2+}$  efflux occurring through a reversal of the  $Ca^{2+}$  uniporter or through activation of an independent efflux pathway should be achieved by a number of different criteria, among which the attainment of a new steady state and the determination of both  $\Delta pH$  and  $\Delta\psi$  play a critical role. On the basis of these criteria evidence for a  $\Delta\psi$ -modulated pathway for  $Ca^{2+}$  efflux, stimulated by  $P_i$  and requiring  $Mg^{2+}$ , is presented in [13].

### Acknowledgements

We are indebted to Professor G. F. Azzone for helpful discussions and for reading the manuscript. The expert technical assistance of Mr Luciano

Pregolato in the isotope experiments is gratefully acknowledged. We thank Mrs Franca Schiavon Mazzari for the excellent typewriting of the manuscript.

### References

- [1] Saris, N. E. and Ackerman, K. E. O. (1980) *Curr. Top. Bioenerg.* 10, 103–179.
- [2] Carafoli, E. (1979) *FEBS Lett.* 104, 1–5.
- [3] Nicholls, D. G. and Crompton, M. (1980) *FEBS Lett.* 111, 261–268.
- [4] Nicholls, D. G. (1978) *Biochem. J.* 176, 463–474.
- [5] Rossi, C. S. and Lehninger, A. L. (1964) *J. Biol. Chem.* 239, 3971–3980.
- [6] Roos, L., Crompton, M. and Carafoli, E. (1980) *Eur. J. Biochem.* 110, 319–325.
- [7] Coehlo, J. L. C. and Vercesi, A. E. (1980) *Arch. Biochem. Biophys.* 204, 141–147.
- [8] Zoccarato, F., Rugolo, M., Siliprandi, D. and Siliprandi, N. (1981) *Eur. J. Biochem.* 114, 195–199.
- [9] Massari, S., Balboni, E. and Azzone, G. F. (1972) *Biochem. Biophys. Acta* 283, 16–22.
- [10] Kobayashi, Y. and Maudsley, D. V. (1969) *Methods Biochem. Anal.* 17, 55–133.
- [11] Addanki, S., Cahill, F. D. and Sotos, J. F. (1968) *J. Biol. Chem.* 243, 2337–2348.
- [12] Becker, G. L. (1980) *Biochim. Biophys. Acta* 591, 234–239.
- [13] Bernardi, P. and Azzone, G. F. (1982) *FEBS Lett.* 139, 13–16.
- [14] Pfeiffer, D. R., Hutson, S. M., Kauffman, R. F. and Lardy, H. A. (1976) *Biochemistry* 15, 2690–2697.
- [15] Lehninger, A. L. (1962) *Physiol. Rev.* 42, 467.
- [16] Chappell, J. B. and Greville, G. D. (1963) *Biochem. Soc. Symp. Cambr.* 23, 39.
- [17] Ernster, L. (1956) *Exp. Cell Res.* 10, 704–720.
- [18] Meisner, H. and Klingenberg, M. (1968) *J. Biol. Chem.* 243, 3631–3639.
- [19] Azzi, A. and Azzone, G. F. (1965) *Biochim. Biophys. Acta* 113, 438–444.
- [20] Pfeiffer, D. R., Schmidt, P. C., Beatrice, M. C. and Schmid, H. H. O. (1979) *J. Biol. Chem.* 254, 11485–11494.